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Meant as a companion to The ACS Style Guide, not a competitor, this book is an extraordinary resource for upper-level chemistry majors as well as graduate students faced with writing a journal article, a conference abstract, or a thesis. Full of prepared research projects and exercises, WriteLike a Chemist provides expert instruction ideal for students from diverse backgrounds, including both native and nonnative speakers of English. It is

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specifically designed to help students transition from the writing skills required in undergraduate lecture and laboratory classes to writing skills required by career chemists: a journal article, a scientific poster, and a research proposal. Each of these types of writing is directed towards a different audience, and writing for a journal requires a different writing style than writing a research proposal for the National Science Foundation. Thus to write like a chemist requires that one learns to write for different audiences. This book assists young scientists in developing that essential writing

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skill.

From models to molecules to mass spectrometry-solve organic chemistry problems with ease Got a grasp on the organic chemistry terms and concepts you need to know, but get lost halfway through a problem or worse yet, not know where to begin? Have no fear - this hands-on guide helps you solve the many types of organic chemistry problems you encounter in a focused, step-by-step manner. With memorization tricks, problem-solving shortcuts, and lots of hands-on practice exercises, you'll sharpen your skills and improve your performance.

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pathway. The outstanding importance of the ubiquitin pathway has been recognized with the 2004 Nobel Prize in Chemistry for Aaron Ciechanover, Avram Hershko, and Irwin Rose. Aaron Ciechanover is one of the editors of this series, and Avram Hershko has contributed to the opening chapter of the present volume. Drawing on the the expertise of two Nobel prize winners, this handy reference compiles information on the initial steps of the ubiquitin pathway. Starting out with a broad view of protein degradation and its functions in cellular regulation, it then goes on to

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examine the molecular mechanisms of ubiquitin conjugation and recycling in detail. All currently known classes of ubiquitin protein ligases are treated here, including latest structural data on these enzymes. Further volumes in the series cover the function of the proteasome, and the roles of the ubiquitin pathway in regulating key cellular processes, as well as its pathophysiological disease states. Required reading for molecular biologists, cell biologists and physiologists with an interest in protein degradation.

Active Learning in General Chemistry

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Searching the Chemical Literature

Organic Chemistry 1

ACS General Chemistry Study Guide

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Green Chemistry - a new approach to designing chemicals and chemical transformations that are beneficial for human health and the environment - is an area that continues to emerge as an important field of study.

Practitioners design to be more sustainable the materials, products, and processes that are the basis of our technologically advanced

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society and economy. Molecular designers are seeing new performance capabilities in the products, new efficiencies in the processes, and achievements in meeting the goals for protecting human health and the environment in a profitable way. Educators have recognized that Green Chemistry principles and practice have not been a part of traditional training in chemistry, and are not part of the skill sets of most practicing chemists. Leaders in Green Chemistry education have developed a wide range of new approaches, courses, tools, and materials that have been introduced and demonstrated in

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the chemistry curriculum in colleges and universities around the U.S. This ACS Symposium Series Book collects the current research and advances in the field of green chemistry, with an emphasis on providing educators with the knowledge and tools needed to incorporate recent information about this field into the chemistry curriculum. This volume is an outstanding resource for any chemical educator wishing to deepen, broaden, or begin the inclusion of green principles and practices into their teaching or research. Given the current interest in green chemistry, this timely book provides an

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Timberlake's Chemistry: An Introduction to General, Organic, and Biological Chemistry is designed to help prepare students for health-related careers, such as nursing, dietetics, respiratory therapy, and environmental or agricultural science. Assuming no prior knowledge of chemistry, it aims to make this

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course an engaging and positive experience by relating the structure and behavior of matter to its role in health and the environment.

Timberlake maintains the clear, friendly writing style and the real-world, health-related applications that have made this text a leader in the discipline. The Eleventh Edition introduces more problem-solving strategies—including new Concept Checks, more Guides to Problem Solving, and more conceptual, challenge, and combined problems.

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*as told by the Periodic Table. Why did Gandhi hate iodine (I, 53)? How did radium (Ra, 88) nearly ruin Marie Curie's reputation? And why is gallium (Ga, 31) the go-to element for laboratory pranksters?**

The Periodic Table is a crowning scientific achievement, but it's also a treasure trove of adventure, betrayal, and obsession. These fascinating tales follow every element on the table as they play out their parts in human history, and in the lives of the (frequently) mad scientists who discovered them. THE DISAPPEARING SPOON masterfully fuses science with the classic lore of

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*invention, investigation, and discovery--from the Big Bang through the end of time. *Though solid at room temperature, gallium is a moldable metal that melts at 84 degrees Fahrenheit. A classic science prank is to mold gallium spoons, serve them with tea, and watch guests recoil as their utensils disappear.*

*Preparing for Your ACS Examination in Organic Chemistry
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The First Forty Years

*ACS Organic Chemistry Exams - the Official Guide
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ACS Style Guide

Structure and Function

Organic Chemistry II For Dummies

Third Chemical Congress of North America,

Toronto, Canada, June 5-10, 1988

Chemistry in the Community

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Changing the Course of Chemistry

A plain-English guide to one of the toughest courses around So, you survived the first semester of Organic Chemistry (maybe even by the skin of your teeth) and now it's time to get back to the classroom and lab! Organic Chemistry II For Dummies is an easy-to-understand reference to this often challenging subject. Thanks to this book, you'll get friendly and comprehensible guidance on everything you can expect to encounter in your Organic Chemistry II course. An extension of the successful Organic Chemistry I For Dummies Covers topics in a straightforward and effective manner Explains concepts and terms in a fast and easy-to-understand way Whether

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become proficient at approaching new situations methodically, based on a repertoire of skills. These skills are vital for successful problem solving in organic chemistry. Existing textbooks provide extensive coverage of the principles but there is far less emphasis on the skills needed to actually solve problems.

A Guide and Resource

Preparing for Your ACS Examination in Physical Chemistry

Translating the Basic Concepts

Study Guide and Student's Solutions Manual for Organic Chemistry

The Official Guide

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Write Like a Chemist

This textbook provides students with a framework for organizing their approach to the course - dispelling the notion that organic chemistry is an overwhelming, shapeless body of facts.

On the cover of this book is a Pacific yew tree, found in the ancient forests of the Pacific Northwest. The bark of the Pacific yew tree produces Taxol, found to be a highly effective drug against ovarian and breast cancer. Taxol blocks mitosis during eukaryotic cell division. The supply of Taxol from the Pacific yew tree is vanishingly small, however. A single 100-year-

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old tree provides only about one dose of the drug (roughly 300 mg). For this reason, as well as the spectacular molecular architecture of Taxol, synthetic organic chemists fiercely undertook efforts to synthesize it. Five total syntheses of Taxol have thus far been reported. Now, a combination of isolation of a related metabolite from European yew needles, and synthesis of Taxol from that intermediate, supply the clinical demand. This case clearly demonstrates the importance of synthesis and the use of organic chemistry. It's just one of the many examples used in the text that will spark the

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interest of students and get them involved in the study of organic chemistry!

The Survival Guide to Organic Chemistry: Bridging the Gap from General Chemistry enables organic chemistry students to bridge the gap between general chemistry and organic chemistry. It makes sense of the myriad of in-depth concepts of organic chemistry, without overwhelming them in the necessary detail often given in a complete organic chemistry text. Here, the topics covered span the entire standard organic chemistry curriculum. The authors describe subjects which require further

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explanation, offer alternate viewpoints for understanding and provide hands-on practical problems and solutions to help master the material. This text ultimately allows students to apply key ideas from their general chemistry curriculum to key concepts in organic chemistry.

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Study Guide 1

Computational chemistry is a means of applying theoretical ideas using computers and a set of techniques for investigating chemical problems within which common questions vary from molecular geometry to the physical properties of substances. Theory and Applications of Computational Chemistry: The First Forty Years is a collection of articles on the emergence of computational chemistry. It shows the enormous breadth of theoretical and computational chemistry today and establishes how theory and computation have become increasingly linked as methodologies

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and technologies have advanced. Written by the pioneers in the field, the book presents historical perspectives and insights into the subject, and addresses new and current methods, as well as problems and applications in theoretical and computational chemistry. Easy to read and packed with personal insights, technical and classical information, this book provides the perfect introduction for graduate students beginning research in this area. It also provides very readable and useful reviews for theoretical chemists. * Written by well-known leading experts * Combines history, personal

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accounts, and theory to explain much of the field of theoretical and computational chemistry * Is the perfect introduction to the field

Active learning methods can provide significant advantages over traditional instructional practices, including improving student engagement and increasing student learning. Focusing on class-level interventions, the chapters in this book showcase evidence-based techniques to encourage active learning in general chemistry. Contributing authors also include approaches to methods that encourage

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productive ways to engage inside and outside of classroom to support students' transition to university. Faculty and administrators considering more effective general chemistry courses will benefit from reading this volume.

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best book I have seen on safety in the undergraduate laboratory." Chemistry World, March 2011 Laboratory Safety for Chemistry Students is uniquely designed to accompany students throughout their four-year undergraduate education and beyond, progressively teaching them the skills and knowledge they need to learn their science and stay safe while working in any lab. This new principles-based approach treats lab safety as a distinct, essential discipline of chemistry, enabling you to instill and sustain a culture of

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safety among students. As students progress through the text, they'll learn about laboratory and chemical hazards, about routes of exposure, about ways to manage these hazards, and about handling common laboratory emergencies. Most importantly, they'll learn that it is very possible to safely use hazardous chemicals in the laboratory by applying safety principles that prevent and minimize exposures. Continuously Reinforces and Builds Safety Knowledge and Safety Culture Each of the book's eight chapters is organized into three tiers of sections, with a

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Laboratory Safety for Chemistry Students is the ideal solution: Each section can be treated as a pre-lab assignment, enabling you to easily incorporate lab safety into all your lab courses without building in additional teaching time. Sections begin with a preview, a quote, and a brief description of a laboratory incident that illustrates the importance of the topic. References at the end of each section guide your students to the latest print and web resources. Students will also find “Chemical Connections” that illustrate how chemical

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principles apply to laboratory safety and “Special Topics” that amplify certain sections by exploring additional, relevant safety issues. Visit the companion site at

<http://userpages.wittenberg.edu/dfinster/LSCS/>.

This book is important because it is the first textbook in an area that has become very popular in recent times. There are around 250 research groups in crystal engineering worldwide today. The subject has been researched for around 40 years but there is still no textbook at the level of senior

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undergraduates and beginning PhD students. This book is expected to fill this gap. The writing style is simple, with an adequate number of exercises and problems, and the diagrams are easy to understand. This book consists major areas of the subject, including organic crystals and co-ordination polymers, and can easily form the basis of a 30 to 40 lecture course for senior undergraduates.

This is the Student Study Guide and Solutions Manual to accompany Organic Chemistry, 3e. Organic Chemistry, 3rd Edition is not merely a

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compilation of principles, but rather, it is a disciplined method of thought and analysis. Success in organic chemistry requires mastery in two core aspects: fundamental concepts and the skills needed to apply those concepts and solve problems. Readers must learn to become proficient at approaching new situations methodically, based on a repertoire of skills. These skills are vital for successful problem solving in organic chemistry. Existing textbooks provide extensive coverage of, the principles, but there is far less emphasis on the skills

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Metal-based drugs are a commercially important sector of the pharmaceutical business, yet most bioinorganic textbooks lack the space to cover comprehensively the subject of metals in medicine. *Uses of Inorganic Chemistry in Medicine* approaches an understanding of the topic in a didactic and systematic manner. The field of inorganic chemistry in medicine may usefully be divided into two main categories - drugs which target metal ions in some form, whether free or protein-bound, and secondly, metal-based drugs where the central metal ion is usually the key feature of the mechanism of action. This latter category can further be subdivided into pharmacodynamic and chemotherapeutic applications, as well as those of imaging. The book summarises the chemical and

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biological studies on clinically used agents of lithium, gold and platinum, as well as highlighting the research on prospective new drugs, including those based on vanadium and manganese. The coverage allows a clear distinction between pharmacodynamic and therapeutic properties of metal-based drugs and focuses not only on those clinical agents in current use, but also on new drugs and uses. This book serves to fill an important niche, bridging bioinorganic and medicinal chemistry and will undoubtedly be of use to senior undergraduates and postgraduates, as well as being an invaluable asset for teachers and researchers in the discipline.